

***m*-Nitrobenzophenone.** Pd(OAc)<sub>2</sub> (0.0022 g, 0.01 mmol, 1 mol.% Pd) was added in an atmosphere of argon to a mixture of Ph<sub>4</sub>BNa (0.0922 g, 0.25 mmol), Na<sub>2</sub>CO<sub>3</sub> (0.159 g, 1.5 mmol), and *m*-NO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>COCl (0.1858 g, 1 mmol) in dry acetone (9 mL), and the mixture was stirred at 20 °C for 2.5 h. Then acetone was evaporated on a rotary evaporator, and the residue was diluted with water (10 mL), extracted with chloroform (3×5 mL), and dried with MgSO<sub>4</sub>. After chloroform was removed on a rotary evaporator, *m*-nitrobenzophenone (0.149 g, 66%) was obtained, m.p. 96–97 °C (cf. Ref. 8: m.p. 95 °C).

Chalcone was obtained similarly from Ph<sub>4</sub>BNa and cinnamoyl chloride (yield 96%).

Thus, the catalytic reactions found occur under very mild conditions and can be used as a new efficient method for the synthesis of nonsymmetric ketones.

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## References

1. N. A. Bumagin, V. V. Bykov, and I. P. Beletskaya, *Izv. Akad. Nauk SSSR, Ser. Khim.*, 1989, 2394 [*Bull. Acad. Sci. USSR, Div. Chem. Sci.*, 1989, **38**, 2206 (Engl. Transl.)].
2. N. A. Bumagin, V. V. Bykov, and I. P. Beletskaya, *Dokl. Akad. Nauk SSSR*, 1990, **315**, 1133 [*Dokl. Chem.*, 1990 (Engl. Transl.)].
3. V. V. Bykov, A. A. Kir'yanov, N. A. Bumagin, and I. P. Beletskaya, *Izv. Akad. Nauk, Ser. Khim.*, 1996, 1582 [*Russ. Chem. Bull.*, 1996, **45**, 1508 (Engl. Transl.)].
4. A. Suzuki, *Pure Appl. Chem.*, 1985, **57**, 1749.
5. N. A. Bumagin, I. G. Bumagina, and I. P. Beletskaya, *Dokl. Akad. Nauk SSSR*, 1984, **274**, 1103 [*Dokl. Chem.*, 1984 (Engl. Transl.)].
6. C. S. Cho, K. Itotani, and S. Uemura, *J. Organomet. Chem.*, 1993, **443**, 253.
7. H. C. Brown and H. L. Young, *J. Org. Chem.*, 1957, **22**, 719.
8. *Dictionary of Organic Compounds*, Eds. G. Heilbron and H. M. Bumbery, London, 1946, **3**, 101.

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## 1-(1-Trimethylsilylcyclopropyl)germatrane as the first representative of cyclopropylgermatranes

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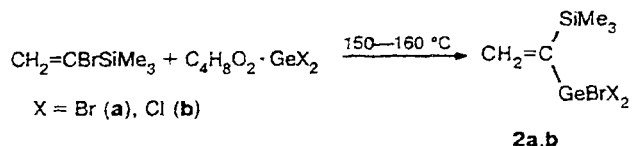
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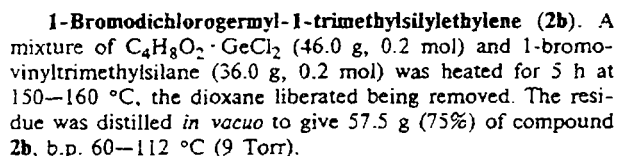
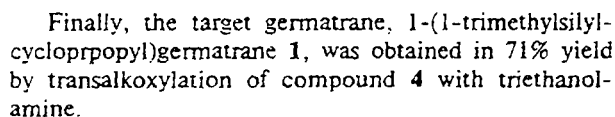
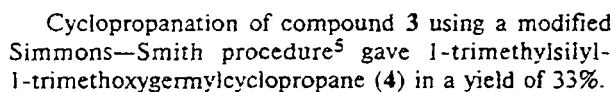
Virtually all germatranes exhibit physiological activities that largely depend on the substituents present in their molecules.<sup>1</sup> Therefore, it seemed of interest to prepare a germatrane molecule containing a cyclopropane fragment, whose presence accounts for the activity of some compounds, in particular, pyrethroids.<sup>2</sup>

In this study, we synthesized the first representative of cyclopropylgermatranes, viz., 1-(1-trimethylsilylcyclopropyl)germatrane (1). As the first step, we prepared 1-tribromogermyl-1-trimethylsilylethylene<sup>3</sup> (2a) (yield 75%) and 1-bromodichlorogermyl-1-trimethylsilylethylene (2b) (yield 82%) by the reactions of 1-bromo-

vinyltrimethylsilane with the dioxane complexes of dichloro- and dibromogermylene, respectively.<sup>4</sup>



Subsequently 1-(trimethylsilyl)vinyltrihalogermenes 2 were converted into 1-trimethylsilyl-1-trimethoxygermylethylene (3) by treating them with methanol.



**1-Trimethylsilyl-1-trimethoxygermylethylene (3).** *A.* 1-Tribromogermyl-1-trimethylsilylethylene (41.1 g, 0.1 mol)<sup>3</sup> was added with stirring to a solution of MeONa (prepared from sodium (6.9 g, 0.3 mol) and methanol (60 mL)); the mixture was stirred for ~1 h at 20 °C, filtered to remove the precipitate, and distilled *in vacuo* to give 16.2 g (60%) of compound 3, b.p. 96–97 °C (19 Torr),  $n_D^{20}$  1.4383.

**B.** Triethylamine (63.0 g, 0.621 mol) was added dropwise with stirring to a solution of compound **2b** (57.5 g, 0.150 mol) in hexane (270 mL) and methanol (20.0 g, 0.621 mol); the

reaction mixture was stirred for 1 h at  $-70^{\circ}\text{C}$ , cooled to  $20^{\circ}\text{C}$ , and filtered from the precipitate. The precipitate was washed with hexane ( $2 \times 50\text{ mL}$ ), the filtrate and the eluent were combined, the solvent was evaporated under atmospheric pressure, and the residue was distilled *in vacuo*. The fraction boiling at  $78\text{--}83^{\circ}\text{C}$  (10 Torr) was collected to give 20.3 g (54%) of product 3,  $n_{\text{D}}^{20}$  1.4430. Repeated distillation gave 15.0 g (38%) of product 3, b.p.  $86.5\text{--}87^{\circ}\text{C}$  (13 Torr),  $n_{\text{D}}^{20}$  1.4446.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$ : 0.03 (s, 9 H,  $\text{SiMe}_3$ ); 3.50 (s, 9 H,  $\text{OCH}_3$ ); 6.38, 6.45 (both d, 2 H,  $\text{CH}_2=\text{}$ ).

**1-Trimethylsilyl-1-trimethoxygermylcyclopropane (4).** Compound **3** (11.0 g, 0.042 mol) and  $\text{CH}_2\text{I}_2$  (20.0 g, 0.075 mol) were added to a freshly prepared zinc-copper pair (5.6 g, 0.100 mol)<sup>5</sup> in ether (40 mL). The reaction mixture was stirred for ~20 h at 50 °C, the ether was evaporated under atmospheric pressure, and the residue was distilled *in vacuo* to give 3.9 g (33%) of compound **4**, b.p. 63–64 °C (1 Torr),  $n_D^{20}$  1.4531. IR,  $\nu/\text{cm}^{-1}$ : 1250, 843 ( $\text{SiMe}_3$ ); 1069, 1045, 850 ( $\text{GeOC}$ ); 912 (cyclopropane ring).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ),  $\delta$ : 0.02 (s, 9 H,  $\text{SiMe}_3$ ); 0.57 (m, 2 H, cyclopropane-ring CH); 0.91 (m, 2 H, cyclopropane-ring CH); 3.52 (s, 9 H,  $\text{OCH}_3$ ).

**1-(1-Trimethylsilylcyclopropyl)germatrane (1).** Triethanolamine (0.7 g, 4.7 mmol) was added to a solution of compound **4** (1.2 g, 4.3 mmol) in benzene (5 mL). The mixture was stirred for ~15 min, and the benzene was evaporated. Recrystallization of the residue (chloroform–hexane, 1 : 1) afforded 1.0 g (71%) of germatrane **1**, m.p. 117–119 °C. IR,  $\nu/\text{cm}^{-1}$ : 3071, 3045, 1097, 1078, 1049, 1020, 914.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ): 8: 0.02 (s, 9 H,  $\text{SiMe}_3$ ); 0.57 (m, 2 H, cyclopropane ring); 0.91 (m, 2 H, cyclopropane ring); 2.80 (t, 6 H,  $\text{CH}_2\text{N}$ ); 3.73 (t, 6 H,  $\text{CH}_2\text{O}$ ).

1. E. Ya. Lukevits, T. K. Gar, L. M. Ignatovich, and V. F. Mironov, *Biologicheskaya aktivnost' soedinenii germaniya* [Biological Activity of Germanium Compounds], Zinatne, Riga, 1990, 191 pp (in Russian).
2. *Piretroidy* [Pyrethroids], Ed. V. K. Promonenkov, Khimiya, Moscow, 1992, 327 pp. (in Russian).
3. N. A. Viktorov, T. K. Gar, I. S. Nikitina, V. M. Nosova, D. A. Ivashchenko, and V. F. Mironov, *Zh. Obshch. Khim.*, 1986, 56, 1535 [*J. Gen. Chem. USSR*, 1986, 56 (Engl. Transl.)].
4. S. P. Kolesnikov, V. I. Shiryayev, and O. M. Nefedov, *Izv. Akad. Nauk SSSR, Ser. Khim.*, 1966, 583 [*Bull. Acad. Sci. USSR, Div. Chem. Sci.*, 1966, 15 (Engl. Transl.)]; V. F. Mironov and T. K. Gar, *Zh. Obshch. Khim.*, 1975, 45, 103 [*J. Gen. Chem. USSR*, 1975, 45 (Engl. Transl.)].
5. V. V. Shcherbinin, Ph. D. (Chem.) Thesis, State Research Institute of Chemistry and Technology of Organoelement Compounds, Moscow, 1975 (in Russian).